## How cottonseed protein concentrate can form a film?

FINAL WORKSHOP OF THE PROJECT COTONBIOMAT
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### PROTEIN-BASED BIODEGRADABLE MATERIAL MANUFACTURE





Wet processing technologies

Low-moisture processing technologies

#### **FILM FORMATION**



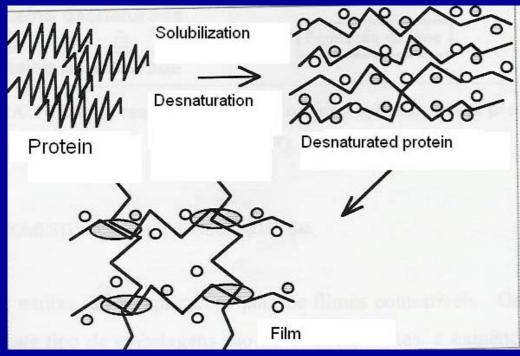
One component able to form a suitably cohesive and continuous matrix



Involved inter- and intra molecular associations or cross-linking of polymer chains

#### **FILM FORMATION**







Fomation a semi-rigid 3D network that entraps and immobilizes the solvents

Font: Cuq, 1993

Degree of cohesion

- -Polymers structure
- -Solvent used
- -Temperature
- -Type of plasticizer
- -Presence of other molecules
- -Degree of crosslinking

#### Formulation of protein based film



Incorporation of a plasticizer to avoid brittleness



Decrease in the molecular forces

Solution: increase of the degree of crosslinking

#### **PROTEIN CONTENT**

#### Flour cottonseed







PB = 32.5 g/100 g of dry matter

#### **COTPROT** solution







PB = 56.3 g/100 g of dry matter

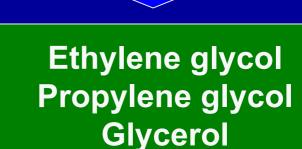
#### **ADDITIVES**

Crosslinking agents

Glutaraldehyde

**Glyoxal** 

**Plasticizers** 

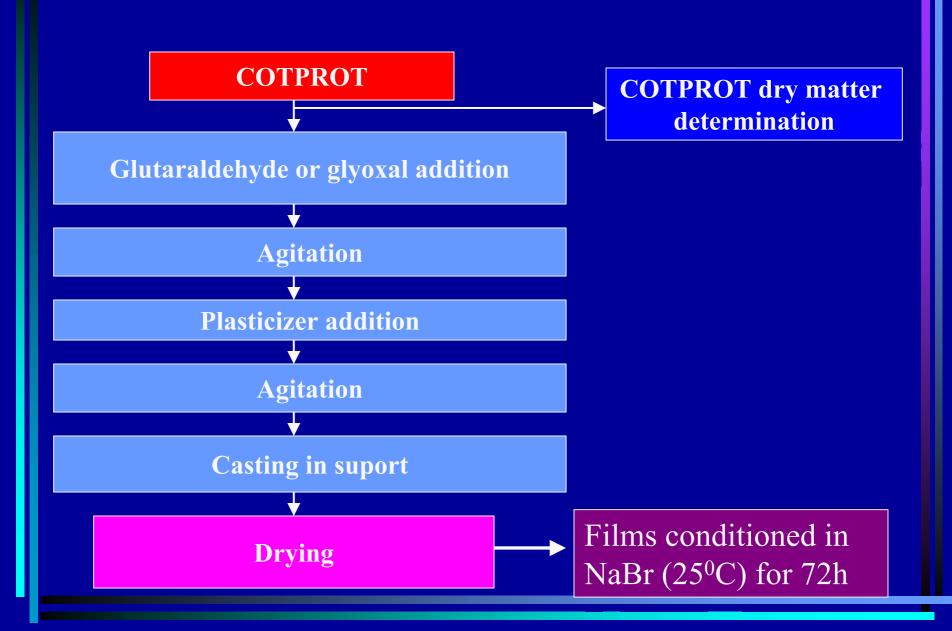


Diethylene glycol
Sorbitol
Polyethylene glycol 300
Polyethylene glycol 400

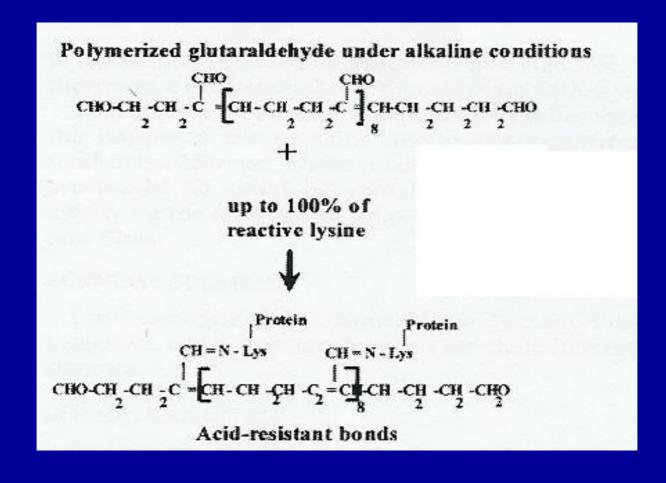
### PLASTICIZERS: MOLAR CONCENTRATIONS AND MOLECULAR MASS

Plasticizer type	Formula	MW	mol plasticizer 100g drymatter	g Oxygen 100g dry matte	mol Oxygen in plasticizer 100g dry matter
Ethylene glycol	$C_2H_6O_2$	62	0.32	10.32	0.166
Prolpylene glycol	$C_3H_8O_2$	76	0.26	8.42	0.111
Glycerol	$C_3H_8O_3$	92	0.22	10.43	0.113
Diethylene glycol	$C_4H_{10}O_3$	106	0.19	9.06	0.085
Sorbitol	$C_6H_{14}O_6$	182	0.11	10.55	0.058
<b>PEG 300</b>	H(OCH <sub>2</sub> - CH <sub>2</sub> ) <sub>6</sub> OH	300	0.07	7.47	0.025
PEG 400	$H(OCH_2 - CH_2)_8 OH$	400	0.05	7.20	0.018

#### **COTTONSEED CROSSLINKED FILMS PRODUCTION**



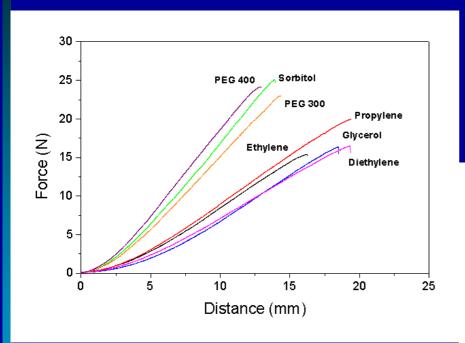
## Postulated mechanism of protein cross-linking by glutarldehyde (Marquié, 2001)



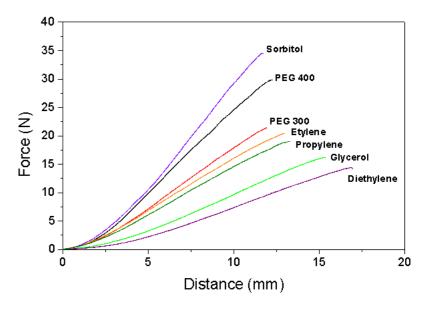
## Postulated mechanism of protein cross-linking by glyoxal (Marquié, 2001)

#### **EXAMPLES OF PUNCTURE CURVES**

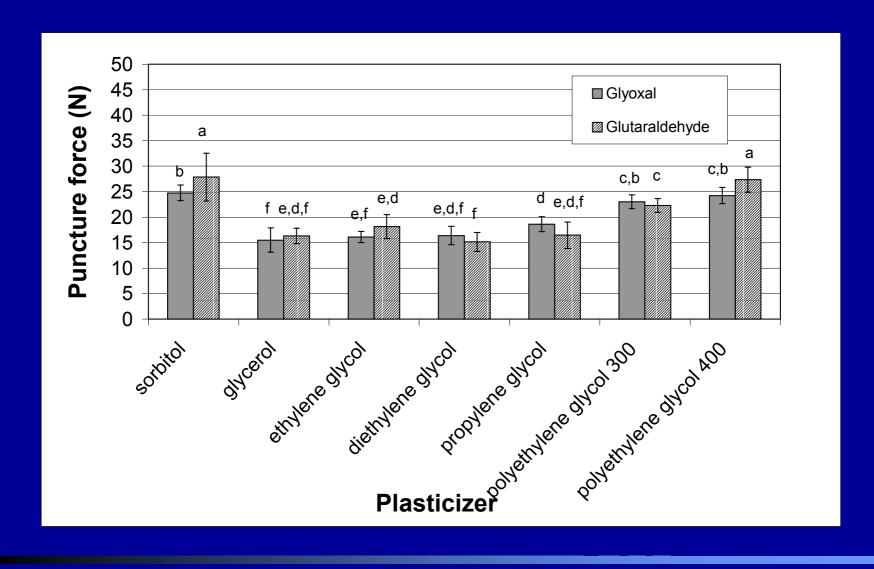
#### Glyoxal



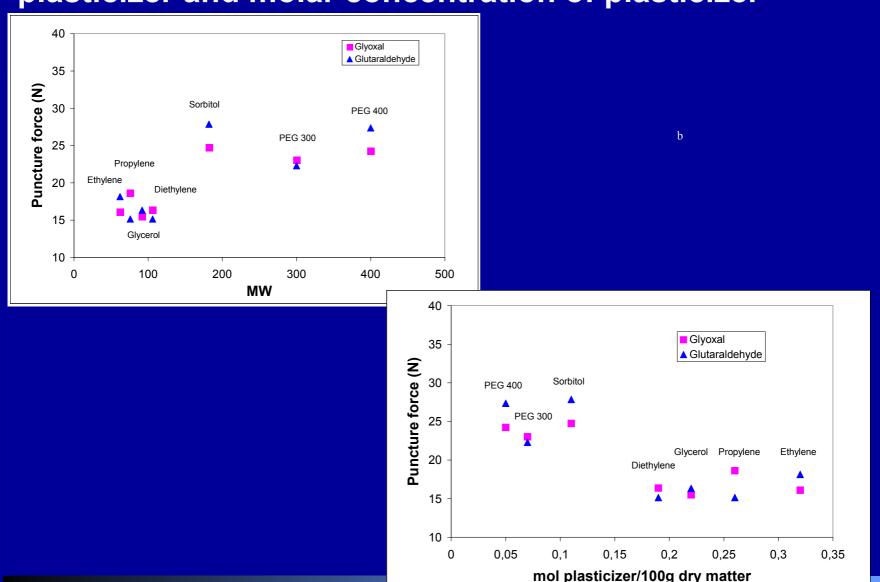
#### Glutaraldehyde



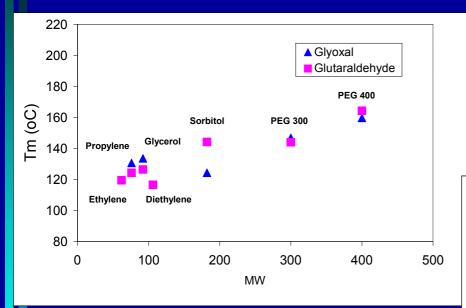
#### **EFFECT OF PLASTICIZER ON PUNCTURE FORCE**

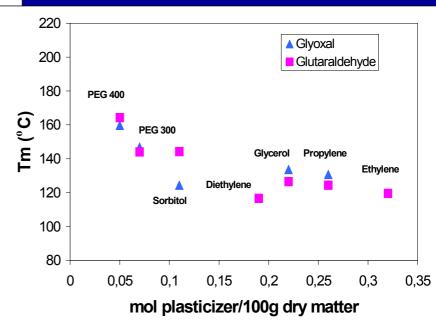


## Puncture force as function of: molecular weight of the plasticizer and molar concentration of plasticizer

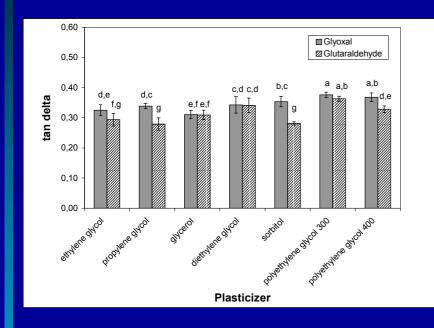


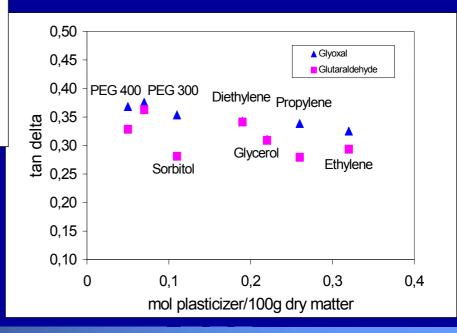
# Melting temperature (Tm) as function of: molecular weight of the plasticizer and molar concentration of plasticizer





## Effect of the type of plasticizer on Loss modulus (MPa) of glutaraldehyde and glyoxal cross-linked cottonseed protein films





#### TRANSGLUTAMINASE CROSSLINKED FILM

#### **Properties**

Puncture force - 4.46±0.56 N -

L\* - 35.3±3.5

Puncture deformation - 12.33±0.99 %

a\* - 31.2±1.4

Tensile strength - 0.27±0.03 Mpa

b\* - 54.6±4.1

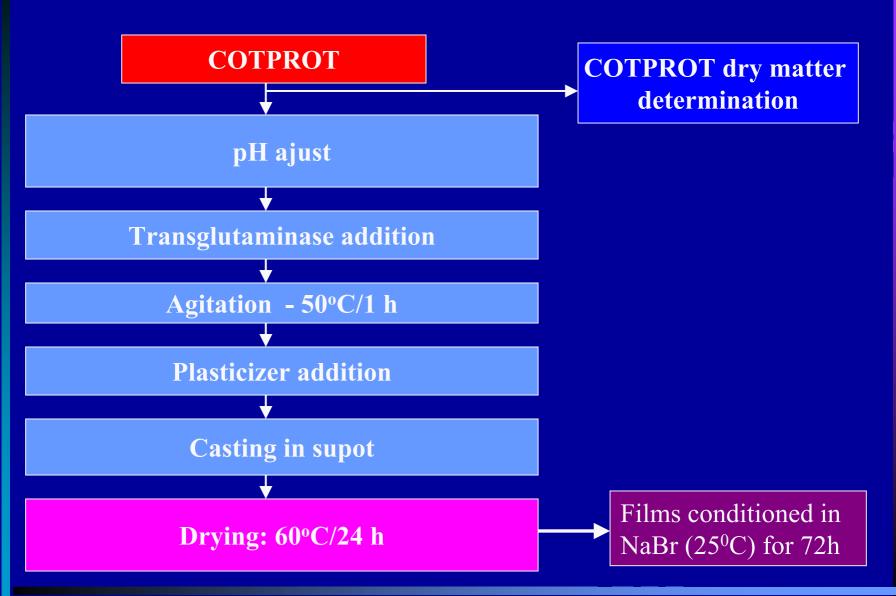
Deformation at break - 100.58±21.80 %

 $\Delta E^* - 86.2 \pm 0.8$ 

Elastic modulus - 0.023±0.007(MPa)

Opacity - 23.4±2.3

#### FILM PRODUCTION USING TRANSGLUTAMINASE



#### **CONSIDERATIONS**

- In relation to the cross-linking agent, an interesting, is the increasing of viscosity during cross-linking observed when gluraraldehyde is used contrary to the glyoxal.
- The viscosity of the filmogenic solution after the crosslinking reaction with transglutaminase did not increased.

## Reactive Lysine Content (by CIRAD) of the Cottonseed Flours

Reactive lysine (g/100g dry matter)				
Repetitions	Average value			
0.92				
0.89				
Solubilization 12h, T = ambient - 0.96				
Solubilization 12h, T = ambient - 1.28	$1.06 \pm 0.17$			
Solubilization 48h, T = ambient - 0.89				
Solubilization 48h, T = ambient - 1.30				
Solubilization 48h, T = 4 °C - 1.08				
Solubilization 48h, T = 4 °C - 1.14				
0.76				
1.26	$1.12 \pm 0.28$			
1.03				
1.41				
0.79				
1.24				
0.98	1.02 ± 0.17			
0.97				
1.12				
	Repetitions  0.92 0.89  Solubilization 12h, $T = ambient - 0.96$ Solubilization 12h, $T = ambient - 1.28$ Solubilization 48h, $T = ambient - 0.89$ Solubilization 48h, $T = ambient - 1.30$ Solubilization 48h, $T = 4$ °C - 1.08 Solubilization 48h, $T = 4$ °C - 1.14  0.76 1.26 1.03 1.41 0.79 1.24 0.98 0.97			



Significant variation on the solution viscosity

